## UNUSUAL TRANSFORMATION IN THE CYCLOADDITION OF AROYLIMINOCHLORO-METHANESULFENYL CHLORIDES TO 3,3-DIMETHYL-2-BUTENE

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Keywords: alkenes, sulfenyl chlorides, heterocyclization.

Fragment R is usually retained in the product structure upon heterocycle formation in the reactions of N-substituted iminochloromethanesulfenyl chlorides RN=CCISCI with alkenes [1-6].

We have found that a sequence of transformations occurs in the reaction of N-benzoyl- (1a) and N-4-chlorobenzoyliminochloromethanesulfenyl (1b) chlorides with 3,3-dimethyl-2-butene (2) in nitromethane in the presence of lithium perchlorate, involving addition of the sulfur-containing electrophile at the double bond, 1,2-shift of methyl group, ring closure with nucleophilic participation of the nitrogen atom of the sulfenyl fragment, and elimination of the aroyl group. Heterocyclization product **3** is formed in 57% yield (in the case of sulfenyl chloride **1a**) and 52% yield (in the case of **1b**).



**2-Chloro-4,4,5-trimethyl-5,6-dihydro-4H-1,3-thiazine (3).** Sample of lithium perchlorate (30 mmol) in nitromethane (40 ml) and solution of alkene **2** (10 mmol) in nitromethane (10 ml) were added with stirring to solution of sulfenyl chloride **1a** or **1b** (10 mmol) in nitromethane (100 ml) at 20°C. After 30 min, the solvent was distilled off in vacuum. The residue was dissolved in methylene chloride (200 ml). The precipitates of LiCl and LiClO<sub>4</sub> were filtered off and washed repeatedly on the filter with methylene chloride. The filtrate was evaporated in vacuum. Recrystallization of the residue from 1:5 hexane–methylene chloride gave compound **3**; mp 122°C (dec.). IR spectrum (vaseline oil), v, cm<sup>-1</sup>: 1565 (C=N). <sup>1</sup>H NMR spectrum (DMSO-d<sub>6</sub>, 300 MHz),  $\delta$ , ppm, *J* (Hz): 3.17 and 2.92 (2H, dd and dd, <sup>3</sup>*J* = 3.8, 8.0, <sup>2</sup>*J* = 12.6, CH<sub>2</sub>S); 1.75 (1H, m, CH); 1.17 and 1.06 (6H, s and s, 2CH<sub>3</sub>); 0.95 (3H, d, <sup>3</sup>*J* = 6.6, 5-CH<sub>3</sub>). Found, %: C 47.20; H 6.73; N 7.81; S 17.92. C<sub>7</sub>H<sub>12</sub>CINS. Calculated, %: C 47.32; H 6.81; N 7.88; S 18.04. Mass spectrum, *m/z* (*I*<sub>rel</sub>, %): 179 [M+2] (4), 177 [M<sup>+</sup>] (12), 135 (20), 100 (100), 41 (35).

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## REFERENCES

- 1. G. Ottmann and H. Hooks, *Angew. Chem.*, **78**, 210 (1966).
- 2. G. Ottmann, H. Hoberecht, and H. Hooks, *Angew. Chem.*, **79**, 1063 (1967).
- 3. E. Kuhle and G. Zumach, Angew. Chem., 82, 63 (1970).
- 4. E. Kuhle, *The Chemistry of the Sulfenic Acids*, Thieme Verlag, Stuttgart (1973), p. 163.
- 5. I. V. Koval', Usp. Khim., 64, 781 (1995).
- 6. A. V. Borisov, I. V. Bodrikov, G. N. Borisova, V. K. Belsky, W. A. Smit, and A. I. Lutsenko, *Mendeleev Commun.*, 52 (1996).